# MACHEREY-NAGEL NUCLEODUR<sup>®</sup> and NUCLEOSHELL<sup>®</sup>



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# Basics

High performance liquid chromatography (HPLC) is part of liquid chromatographic separating processes of substance mixtures and their analysis. In the beginning the technique was also called high pressure liquid chromatography due to the high back pressure of the column. HPLC offers qualitative (identification of substances) and quantitative (concentration determination) analysis by comparison with standard substances. The term HPLC was introduced in the 1970s to describe the high performance method developed from the column liquid chromatography that came about in the 1930s. At the beginning of the 21st century HPLC was complemented by even more efficient UHPLC (ultra high performance liquid chromatography). Hereby even higher pressures (> 400 bar) result in shorter analysis time and enhanced efficiency enabling a higher sample throughput with smaller sample volumes.

### Application

HPLC/UHPLC is used additionally to gas chromatography (GC) for separation and determination of complex substance mixtures composed of low-volatile, polar and ionic, high-molecular or thermal instable substances. Therefore, a sufficient solubility of the sample in a solvent or a solvent mixture is required. HPLC/UHPLC is used for purity control of chemicals and industrial products, determination of active agents for drug development, production and testing, environmental analytics, quality and purity control of foods, analysis of ingredients in cosmetics as well as isolation of biopolymers.

### Basic principle

In liquid column chromatography a mobile phase (eluent) flows through a particle filled tube (separation column, stationary phase). In classic column chromatography this tube is a glass column with an inner diameter of several centimeters and a length up to 450 mm or even bigger. The filling material typically consists of coarse-grained particles like silica gel 60. The eluent is transported through the separation column either by hydrostatic pressure or a low-pressure pump with 1.5–2 bar.

In contrast HPLC columns consist of stainless steel with an inner diameter of 2–4.6 mm and a length of 20–300 mm. The column packing, mostly modified porous silica, has generally a particle size of 3, 5, 7 or 10  $\mu$ m and a pore size of 50, 100, 120 (for low-molecular analytes) or 300–4000 Å (for high-molecular analytes). In UHPLC shorter columns in the range of 20–150 mm length with highly efficient particles of 1.8  $\mu$ m size (sub-2  $\mu$ m) are utilized. A guard column of a few millimeters length can be utilized and installed with a specific Column Protection System to increase the column lifetime. HPLC/UHPLC uses a high-pressure pump to transport the eluent from a storage vessel into the system with a column back pressure of up to 600/1200 bar.

# **Basics**

### Instrument

HPLC as well as UHPLC instruments have different building blocks. The storage vessel (eluent reservoir, 1) usually contains a deaerator unit (3) for the solvents. Followed by a gradient valve (4) with mixing chamber (5) in flow direction, which allows the usage of isocratic as well as gradient methods. A high-pressure pump (6) transports the sample into the system. The sample is injected via an injection valve (7). Usually this is operated automatically with a syringe by an autosampler. With the eluent flow the sample is transported to guard and separating column (8). For better reproducibility of the separation tempering with a column oven (2) should be performed. The separated substances are determined with a detector (9). In the resulting chromatogram each detector signal of a substance (peak), is related to the retention time of the column. With the data evaluation (10) these peaks can be identified and their concentration can be determined.



### Separation mechanism

While flowing through the column each component of the solved mixture interacts differently with the stationary phase. According to the characteristics of the substance (hydrophobic, polar, ionic, aromatic, sterically hindered etc.) the strength of the interactions vary and thus the compounds are retained by the stationary phase in different ways. Essentially a distinction is drawn between normal phase (NP), reversed phase (RP) and ion exchange chromatography. Depending on the structure of the stationary phase diverse interactions e.g., van der Waals forces or  $\pi$ - $\pi$ -stacking can occur and different polar mobile phases are required. For polar stationary normal phases (e.g., SiOH, CN, OH, NH<sub>2</sub>) non-polar eluents like *n*-heptane, hexane, dichloromethane or 2-propanol are applicable. While for reversed phases (e.g., C<sub>18</sub>, C<sub>8</sub>, C<sub>4</sub>, C<sub>2</sub>, C<sub>6</sub>H<sub>5</sub>) typically polar RP eluents (e.g., acetonitrile or methanol with ultrapure water or buffer) and for ion exchange (e.g., SA, SB) aqueous buffers (e.g., phosphate, acetate, citric buffer) come to use.

# **Basics**

### Selectivity

The characteristic separation behavior of phases under certain conditions is also called selectivity. This is dependent on different parameters like structure and modifications of the base silica gel, nature of the chemical binding or the type of endcapping.

In recent decades several methods have been developed to compare and distinguish the selectivity of various silica gels and their modifications. In this connection defined substances or substance classes are analyzed and the chromatographic parameters are graphically presented. A frequently applied model in specialist literature is e.g., the TANAKA plot, which allows a quick comparison of different HPLC phases [1].



hydrogen bonding capacity

Parameter of the Tanaka diagram:

Capacity = k' (pentylbenzene) Hydrophobicity =  $\alpha$  (pentylbenzene, butylbenzene) Steric selectivity =  $\alpha$  (triphenyl, o-terphenyl)

Hydrogen bonding capacity (capacity of silanol) = α (caffeine, phenol) lon exchange capacity at pH 2.7 = α (benzylamine, phenol) lon exchange capacity at pH 7.6 =  $\alpha$  (benzylamine, phenol)

The comparison of NUCLEOSHELL® RP18 and NUCLEOSHELL RP® 18plus for example shows a lower ion exchange capacity at pH 7.6 for the monomeric NUCLEOSHELL® RP 18plus. The radar chart also reflects a more pronounced steric selectivity of NUCLEOSHELL® RP 18 due to a higher density of modifications with C18 chains.

### Characteristic parameters

The success of a chromatographic separation depends apart from the stationary and mobile phase also on other characteristics like the quality of the separating column or the linear flow rate. The following schematic chromatogram illustrates the most important parameters which characterize a separation.



Schematic chromatogram legend							
Peak width:	Peak width:						
W <sub>1/2</sub>	peak width at half height						
W	peak width of the peak (intersection point of the inflectional tangents with the zero line)						
Peak symme	Peak symmetry:						
А	peak front to peak maximum at 10% of peak height						
В	peak maximum to peak end at 10 % of peak height						
Retention tin	ne:						
t <sub>o</sub>	dead time of a column = retention time of a non-retarded substance						
t <sub>R1</sub> , t <sub>R2</sub>	retention times of components 1 and 2						
ť <sub>R1</sub> , ť <sub>R2</sub>	net retention times of components 1 and 2						

In a chromatographic system the substances differ from each other in their retention time in or on the stationary phase. The time, which is needed by a sample component to migrate from column inlet (sample injection) to the column end (detector) is the retention time  $t_{R1}$  or  $t_{R2}$ . The dead time  $t_0$  is the time required by an inert compound to migrate from column inlet to column end without any retardation by the stationary phase. Consequently, the dead time is identical with the retention time of the sample component remaining in the stationary phase. The difference of total retention time and dead time yields the net retention time  $t'_{R1}$  or  $t'_{R2}$ , which is the time a sample component remains in the stationary phase.

$$t'_{R1} = t_{R1} - t_0$$
 bzw.  $t'_{R2} = t_{R2} - t_0$ 

To compare chromatograms that are recorded with columns of different lengths and internal diameters, as well as different flow rates, the retention time is converted into a dimensionless capacity factor k'.

$$k'_{1} = \frac{t_{R1} - t_{0}}{t_{0}} \quad \text{bzw.} \quad k'_{2} = \frac{t_{R2} - t_{0}}{t_{0}}$$

The relative retention  $\alpha$ , also known as the separation factor, describes the ability of a chromatographic system (stationary and mobile phase) to distinguish between two compounds. This is calculated from the rate of the capacity factors of the substances, where the figure in the denominator is the reference compound.

$$\alpha = \frac{k'_2}{k'_1}$$

The resolution R is a measure for the efficiency of the column to separate two substances. Besides the retention time  $t_{\rm R}$  the peak width at half height  $w_{\rm 1/2}$  is also included.

$$R = 1.18 \cdot \frac{t_{R2} - t_{R1}}{(w_{1/2})_2 + (w_{1/2})_1}$$

For practical reasons the peak symmetry is calculated at 10 % of peak height. Ideally symmetry should be 1, i.e. A = B. Values > 1 indicate peak tailing, while values < 1 indicate peak fronting.

Peak symmetry = 
$$\frac{B}{A}$$

Instead of the mobile phase volumetric flow rate [mL/min], which is controlled at the HPLC instrument, it is advantageous to use the linear velocity u [cm/sec]. The linear velocity is independent of the column cross section and proportional to the pressure drop in the column. The linear velocity can be calculated by means of the dead time, where L is the column length in cm and  $t_0$  the dead time in sec.

$$u = \frac{L}{t_0}$$

The quality of a column packing is determined through the number of theoretical plates N. High N values indicate a high capability to separate complex sample mixtures.

$$\mathsf{N} = 5.54 \cdot \left(\frac{\mathsf{t}_{\mathsf{R1}}}{\mathsf{w}_{\mathsf{1/2}}}\right)^2$$

The value of the height equivalent to a theoretical plate HEPT is a criterion for the quality of a column. HEPT, is the length, in which the chromatographic equilibrium between mobile and stationary phase has been adjusted once. Its value depends on the particle size, the flow velocity, the mobile phase viscosity and especially on the packing quality. Small HEPT values, meaning a large number of theoretical plates N, facilitate the column to separate complex sample mixtures.

$$H = \frac{L}{N}$$

The Van Deemter equation shows the dependence of the HEPT on the velocity u.



A term = eddy-diffusion, B term = longitudinal diffusion coefficient,

C term = mass transfer coefficient, H = HEPT = height equivalent to a theoretical plate

The A term, also called eddy-diffusion, is a function of the particle size, the B term a function of the diffusion coefficient of the substance in the mobile phase and the C term the retardation of a substance by the interface between stationary and mobile phase. At the point of intersection of  $h_{min}$  and  $u_{opt}$  the optimal separation efficiency for a column with high peak symmetry for the separated substances is obtained.

# Column quality

Each HPLC/UHPLC column of MACHEREY-NAGEL is individually tested according to the most important characteristic parameters in quality control and the results are documented in a certificate of analysis.

Detailed information of the particular properties of the modern high-purity silica phases NUCLEODUR<sup>®</sup> and Core-Shell material NUCLEOSHELL<sup>®</sup> as well as the respective HPLC- and UHPLC-columns can be found on the following pages.



# USP listing

USP specif	USP specification of MN HPLC phases						
Code	Specification	MN HPLC Phases	Page				
USP L1	octadecyl silane chemically bonded to porous silica particles 1.5 to 10 $\mu m$ diameter, or	NUCLEODUR <sup>®</sup> C18 ec	38				
	monolithic silica gel	NUCLEODUR <sup>®</sup> C18 Gravity	18				
		NUCLEODUR <sup>®</sup> C18 Gravity-SB	22				
		NUCLEODUR <sup>®</sup> C18 HTec	45				
		NUCLEODUR <sup>®</sup> C18 Isis	24				
		NUCLEODUR® C18 Pyramid	26				
		NUCLEODUR <sup>®</sup> PolarTec	28				
		NUCLEODUR <sup>®</sup> Sphinx RP	36				
		NUCLEOSHELL <sup>®</sup> RP 18	70				
		NUCLEOSHELL <sup>®</sup> RP 18plus	72				
		NUCLEOSHELL Bluebird RP 18	75				
USP L3	porous silica particles, 1.5 to 10 $\mu m$ diameter, or monolithic silica gel	NUCLEODUR <sup>®</sup> SIOH	54				
USP L7	octyl silane chemically bonded to totally porous silica particles, 1.8 to 10 $\mu m$ diameter	NUCLEODUR <sup>®</sup> C8 ec	38				
		NUCLEODUR <sup>®</sup> C8 Gravity	18				
USP L8	an essentially monomolecular layer of aminopropyl silane chemically bonded to totally porous silica gel support, 1.5 to 10 $\mu m$ diameter	NUCLEODUR <sup>®</sup> NH2/NH2-RP	52				
USP L10	nitrile groups chemically bonded to porous silica particles, 1.5 to 10 $\mu m$ diameter	NUCLEODUR <sup>®</sup> CN/CN-RP	50				
USP L11	phenyl groups chemically bonded to porous silica particles, 1.5 to 10 $\mu m$ diameter	NUCLEODUR <sup>®</sup> Phenyl-Hexyl	30				
		NUCLEODUR <sup>®</sup> π <sup>2</sup>	34				
		NUCLEODUR <sup>®</sup> Sphinx RP	36				
		NUCLEOSHELL <sup>®</sup> Phenyl-Hexyl	78				
		NUCLEOSHELL Biphenyl	81				
USP L26	butyl silane chemically bonded to totally porous silica particles, 5 to 10 $\mu m$ diameter	NUCLEODUR <sup>®</sup> C4 ec	38				
USP L43	pentafluorophenyl groups chemically bonded to silica particles by a propyl spacer, 1.5 to 10 $\mu m$	NUCLEODUR <sup>®</sup> PFP	32				
	diameter	NUCLEOSHELL <sup>®</sup> PFP	84				
USP L60	spherical porous silica gel, particle size of 10 $\mu m$ diameter or smaller, the surface of which has been covalently modified with alkyl amide groups and endcapped	NUCLEODUR <sup>®</sup> PolarTec	28				
USP L118	aqueous polymerized $C_{\rm 18}$ groups on silica particles, 1.2 to 5 $\mu m$ in diameter	NUCLEODUR <sup>®</sup> C18 PAH	56				

NUCLEODUR<sup>®</sup> is a fully synthetical type B silica (silica of 3rd generation) offering highly advanced physical properties like totally spherical particle shape, outstanding surface microstructure, high pressure stability and low metal content.

NUCLEODUR<sup>®</sup> as a state-of-the-art silica is the ideal base material for modern HPLC phases. It is the result of MACHEREY-NAGEL's pioneering research in chromatography for more than 40 years.

In RP liquid chromatography the efficiency of the packing is strongly affected by the quality of the base silica itself. Shortcomings in the surface geometry of the particles or metal contaminants are the main reasons for inadequate coverage with the covalently bonded alkylsilanes in the subsequent derivatization steps. It is well known, that poor surface coverage and, in consequence, high activity of residual free silanols often results in peak tailing or adsorption, particularly with basic compounds.

### Particle shape and surface symmetry



NUCLEODUR<sup>®</sup> silicas are synthesized in a unique and carefully controlled manufacturing process which provides silica particles, which are totally spherical. The picture shows the outstanding smoothness of the NUCLEODUR<sup>®</sup> surface.

### Purity

As already mentioned above, a highly pure silica is required for achieving symmetric peak shapes and maximum resolution. Inclusions of, e.g., iron or alkaline earth metal ions on the silica surface are largely responsible for the unwanted interactions with ionizable analytes, e.g., amines or phenolic compounds.

NUCLEODUR<sup>®</sup> is virtually free of metal impurities and low acidic surface silanols. Elemental analysis data of NUCLEODUR<sup>®</sup> 5  $\mu$ m measured by AAS are listed on the following page.

Elementary analysis (metal ions) of NUCLEODUR <sup>®</sup> 100-5				
Aluminum	< 5	ppm		
Iron	< 5	ppm		
Sodium	< 5	ppm		
Calcium	< 10	ppm		
Titanium	< 1	ppm		
Zirconium	< 1	ppm		
Arsenic	< 0.5	ppm		
Mercury	< 0.05	ppm		

### Pressure stability

The totally spherical and 100 % synthetic silica gel exhibits an outstanding mechanical stability, even at high pressures and elevated eluent flow rates. In addition, after several cycles of repeated packing, no significant drop in pressure can be observed. The latter is of prime importance for preparative and process-scale applications.

 $\rm NUCLEODUR^{\$}$  silica is available with two pore sizes – 110 Å pore size as standard material and as 300 Å widepore material for the separation of biomolecules, like peptides and proteins.

Physical data of NUCLEODUR <sup>®</sup>						
	Standard	Widepore				
Pore size	110 Å	300 Å				
Surface area (BET)	340 m²/g	100 m²/g				
Pore volume	0.9 mL/g	0.9 mL/g				
Density	0.47 g/mL	0.47 g/mL				

### NUCLEODUR<sup>®</sup> modifications

Several different surface modifications based on NUCLEODUR<sup>®</sup> silica have been developed over the last two decades providing a full range of specified HPLC phases and an ideal tool for every separation.

For a summary of important properties of our NUCLEODUR® phases see page 14.

# CHROMABOND<sup>®</sup> QuEChERS Mixes for sample preparation



# "Quick Easy Cheap Effective Rugged Safe"

- High throughput due to easy handling
- Solvent and time-saving procedure
- High reproducibility and recovery rates
- Broad range of applications (e.g. pesticides from food)



# 1.8 µm particles for increased separation efficiency

### Advantages of 1.8 µm particle size

Miniaturization started in the early stage of HPLC with the reduction of particle size from 10  $\mu$ m via 7  $\mu$ m to standard 5  $\mu$ m – still the most used particle diameter in analytical HPLC – to 3  $\mu$ m spherical particles. With the introduction of 1.8  $\mu$ m NUCLEODUR® particles researchers have turned over a new leaf in HPLC column technology, featuring extraordinary improvements in terms of plate numbers, column efficiency and resolution compared with 3  $\mu$ m particles.

Increased separation efficiency by higher number of theoretical plates (N):

- $\cdot$  50 × 4.6 mm NUCLEODUR<sup>®</sup> C18 Gravity
- $\cdot$  3 µm: N  $\geq$  100 000 plates/m (h-value  $\leq$  10)
- · 1.8  $\mu$ m: N  $\geq$  166 667 plates/m (h-value  $\leq$  6)

Increasing the plate number by ~ 67 % offers the possibility of using shorter columns with equal plate number, therefore resulting in a decrease of analysis time.

Significant improvement in resolution

$$R_{s} = \frac{\sqrt{N}}{4} \left(\frac{\alpha - 1}{\alpha}\right) \left(\frac{k'_{i}}{k'_{i} + 1}\right)$$

$$\label{eq:Rs} \begin{split} R_s = resolution, \, \alpha = selectivity (separation factor), \, k_i^{\, \prime} = retention \\ N = plate number with \, N \propto 1/d_P, \, d_P = particle \, diameter \end{split}$$

### Key features

- Decrease of analysis time (ultra-fast HPLC)
- Shorter columns with high separation efficiency and significant improvement of resolution and detection sensitivity
- Suitable for LC/MS due to low bleeding characteristics

#### Fractionation

 NUCLEODUR<sup>®</sup> 1.8 µm particles are specially fractionated to limit the increase in back pressure.

### Availability

 The following NUCLEODUR<sup>®</sup> phases are available in 1.8 µm: C18 Gravity, C8 Gravity, C18 Gravity-SB, C18 Isis, C18 Pyramid, PolarTec, Phenyl-Hexyl, PFP, Sphinx RP, C18 HTec and HILIC



Use of 1.8  $\mu$ m instead of 3  $\mu$ m particles leads to an increase of resolution by a factor of 1.29 (29 %) since the resolution is inversely proportional to the square root of the particle size.

### Column back pressure

Due to the smaller particles the back pressure will increase according to

$$\Delta_{\rm P} = \frac{\Phi \cdot L_{\rm C} \cdot \eta \cdot u}{d_{\rm P}^2}$$

$$\begin{split} \Delta_{P} = \text{pressure drop}, \ \Phi = \text{flow resistance (non-dimensional)}, \ LC = \text{column} \\ \text{length}, \ \eta = \text{viscosity}, \ u = \text{linear velocity}, \ d_{P} = \text{particle diameter} \end{split}$$

The high sphericity of the NUCLEODUR® particles and a very narrow particle size distribution allow to keep the back pressure on a moderate level.

### Comparison of back pressures

Eluent 100 % methanol, flow rate 1.5 mL/min temperature 22 °C, column dimensions 50 × 4.6 mm

	,	
	NUCLEODUR <sup>®</sup> C18 Gravity	Competitor
3 µm	70 bar	-
1.8 µm	130 bar	170 bar

#### Higher flow rates and shorter run times

The optimal flow rate for 1.8  $\mu m$  particles is higher than for 3 and 5  $\mu m$  particles (see figure – the flow rate should be at the van Deemter minimum).

### Van Deemter curves

#### Van Deemter curves



Column 50 x 4.6 mm, acetonitrile - water (50:50, v/v), analyte toluene

### Technical requirements

To gain best results with 1.8  $\mu$ m particles certain technical demands must be met including pumps for flow rates of 2–3 mL with pressures of 250–1000 bar, minimized dead volume, and fast data recording.



Phase	Specification	Page	Characteristic*	Stability	Structu	re
	octadecyl, high density coating, multi-endcapping 18 % C	18	A●●●● B <b>〔</b>	pH 1–11, suitable for LC/MS	DUR <sup>®</sup> שמר	
	USP L1		COOL		CLEC (Si-C	
C18 Gravity					NN	
	octadecyl (monomeric), extensive endcapping	22		pH 1–9, suitable for LC/MS	® H	· · · -=
	13 % C		C -		EODI I-O <sub>2</sub> ),	EODUR SWIT- X
C18 Gravity-SB	USP L1				(Si (Si	
	octyl, high density coating, multi-endcapping	18	A • • •	pH 1–11, suitable for LC/MS	© Ľ	~~~~
	11 % C		B		0DU 02),	
	USP L7				ICLE (Si-(	
C8 Gravity					N	~~~~
	octadecyl phase with specially crosslinked surface modification, endcapping	24		pH 1–10, suitable for LC/MS	ODUR <sup>®</sup> J <sub>2</sub> ),	
	20 % C				ICLE (Si-(	
C18 Isis	USP L1				NN	
	octadecyl with polar endcapping	26	A	stable in 100 % aqueous	©_	
	14 % C		В ● ● ●	pH 1–9,	DUF ),	
	USP L1		C	suitable for LC/MS	SI-O	
C18 Pyramid					NUO NUO	≥ 0 ~ 0H
	octadecyl with embedded polar	28	A • • • •	stable in 100 % aqueous	e	
	group, endcapping		В●●●	eluent, pH 1–9,	DUR	
	USP 1 1 and 1 60		C	suitable for LC/MS		TMS
PolarTec					NUCI (8	ž ne
	phenylhexyl, multi-endcapping	30	A●●	pH 1–10,	@	
	10 % C		В ● ● ●	suitable for LC/MS	DUR ),	
	USP L11		C		si-o	
Phenyl-Hexyl					WNC (*	
	biphenylpropyl, multi-endcapping	34	A●●€	pH 3–10	© ©	
	17 % C		B		a)n bUF	
	USP L11		C		SI-O	
π <sup>2</sup>					UNU NUC	ž ~~~~
	pentafluorophenylpropyl, multi-endcapping	32	A	pH 1–9, suitable for LC/MS	© ſr	F.
	8% C		B		חסכ מוווע	
	USP L43				CLE( (Si-C	
PFP					Ň	

\* A =  $\bigcirc$  hydrophobic selectivity, B =  $\bigcirc$  polar/ionic selectivity, C =  $\bigcirc$  steric selectivity \*\* phases which provide a similar selectivity based on chemical and physical properties

Application	Similar phases**	Interactions · retention med	chanism
in general compounds with ionizable functional groups such as basic pharmaceuticals and pesticides	NUCLEOSIL <sup>®</sup> C18 HD Xterra <sup>®</sup> RP18 / MS C18; Luna <sup>®</sup> C18(2), Gemini <sup>®</sup> , Synergi <sup>®</sup> Max RP; Zorbax <sup>®</sup> Extend-C18; Inertsil <sup>®</sup> ODS III; Purospher <sup>®</sup> STAR RP-18; Hypersil <sup>™</sup> BDS	hydrophobic (van der Waals interactions)	
overall sophisticated analytical separations, especially for polar compounds, e.g., antibiotics, water-soluble vitamins, organic acids	_	hydrophobic (van der Waals interactions) with additional polar interactions	SI-O-SI(CH3)3 H1C-H3
like C <sub>18</sub> Gravity, however, generally shorter retention times for nonpolar compounds	NUCLEOSIL® C8 HD Xterra® RP8 / MS C <sub>8</sub> ; Luna® C <sub>8</sub> ; Zorbax® Eclipse XDB-C <sub>8</sub>	hydrophobic (van der Waals interactions)	OH SI(CH <sub>3</sub> ) <sub>3</sub> OH OH CH <sub>3</sub> CH <sub>3</sub>
high steric selectivity, thus suited for separation of positional and structural isomers, planar / nonplanar molecules	NUCLEOSIL <sup>®</sup> C18 AB Inertsil <sup>®</sup> ODS-P; Pro C18 RS	steric and hydrophobic	
basic pharmaceuticals, very polar compounds, organic acids	Aqua, Synergi <sup>®</sup> Hydro-RP; AQ; Atlantis <sup>®</sup> dC18; Polaris <sup>®</sup> C18-A	hydrophobic and polar (H bonds)	
basic pharmaceuticals, organic acids, pesticides, amino acids, water-soluble vitamins	NUCLEOSIL <sup>®</sup> C18 Nautilus ProntoSIL <sup>®</sup> C18 AQ, Zorbax <sup>®</sup> Bonus-RP, Polaris <sup>®</sup> Amide-C18; Ascentis <sup>®</sup> RP Amide, SymmetryShield <sup>™</sup> RP18; SUPELCOSIL <sup>™</sup> LC-ABZ+; HyPURITY <sup>™</sup> ADVANCE; ACCLAIM Polar AD.II	hydrophobic and polar (H bonds)	HO SI(CH3)s HO Pol
aromatic and unsaturated compounds, polar compounds like pharmaceuticals, antibiotics	Luna <sup>®</sup> Phenyl-Hexyl; Zorbax <sup>®</sup> Eclipse Plus Phenyl-Hexyl; Kromasil <sup>®</sup> Phenyl-Hexyl	π-π and hydrophobic	O <sub>2</sub> N
aromatic and unsaturated compounds, polar compounds like pharmaceuticals, antibiotics	Pinnacle <sup>®</sup> DB Biphenyl; Ultra Biphenyl	π-π and hydrophobic	
aromatic and unsaturated compounds, halogen compounds, phenols, isomers, polar pharmaceuticals, antibiotics	ACQUITY <sup>®</sup> CSH Fluoro-Phenyl; Hypersil™ GOLD PFP; Luna <sup>®</sup> PFP(2); Discovery <sup>®</sup> HS F5; Allure <sup>®</sup> PFP Propyl; Ultra II PFP Propyl	polar (H bond), dipole-dipole, π-π and hydrophobic	

Phase	Specification	Page	Characteristic*	Stability	Structu	re
Sphinx RP	bifunctional, balanced ratio of propylphenyl and octadecyl, endcapping 15 % C USP L1 and L11	36	A • • • • B • • • • C •	pH 1–10, suitable for LC/MS	NUCLEODUR <sup>®</sup> (Si-O <sub>2</sub> ) <sub>n</sub>	
C18 ec	octadecyl, medium density, endcapping, available in 110 Å and 300 Å pore size 17.5 % / 4 % C USP L1	38	A ● ● ● ● B ● C ● ● ● ●	pH 1–9	NUCLEODUR <sup>®</sup> (Si-O <sub>2</sub> ) <sub>n</sub>	PO-ER-
C8 ec	octyl, medium density, endcapping 10.5 % C USP L7	38	A ● ● B ● ¶ C ● ● ¶	pH 1–9	NUCLEODUR <sup>®</sup> (Si-O <sub>2</sub> ),	
C4 ec	butyl, medium density, endcapping, 300 Å pore size 2.5 % C USP L26	38	A ● B ● ● C ● ¶	рН 1–9	NUCLEODUR <sup>®</sup> (Si-O <sub>2</sub> ),	
C18 HTec	octadecyl, high density coating, high capacity, multi-endcapping 18 % C USP L1	45	A • • • • • • B • C • • •	pH 1–11, suitable for LC/MS	NUCLEODUR <sup>®</sup> (Si-O <sub>2</sub> ) <sub>n</sub>	MICLEOUR
HILIC	zwitterionic ammonium-sulfonic acid phase, no endcapping 7 % C	48	A • B • • • • • C -	pH 2-8.5	NUCLEODUR <sup>®</sup> (Si-O <sub>2</sub> ) <sub>n</sub>	е -S-OH -S-OH -S-OH -S-OH
CN/CN-RP	cyano (nitrile) for NP and RP separations, endcapping 7 % C USP L10	50	A • B • • • • C -	pH 1–8, stable towards highly aqueous mobile phases	NUCLEODUR <sup>®</sup> (Si-O <sub>2</sub> ) <sub>n</sub>	dH-on on ON
NH2/NH2-RP	aminopropyl for NP and RP separations, no endcapping 2.5 % C USP L8	52	A ● B ● ● ● ● C -	pH 2–8, stable towards highly aqueous mobile phases	NUCLEODUR <sup>®</sup> (Si-O <sub>2</sub> ),	det - NH- -si-OH NH- NH- NH-
SIOH	unmodified high purity silica, no endcapping USP L3	54	A - B - C -	pH 2–8	NUCLEODUR <sup>®</sup> (Si-O <sub>2</sub> ) <sub>n</sub>	-9-0H -9-0H -9-0H

\* A =  $\bigcirc$  hydrophobic selectivity, B =  $\bigcirc$  polar/ionic selectivity, C =  $\bigcirc$  steric selectivity \*\* phases which provide a similar selectivity based on chemical and physical properties

Application	Similar phases**	Interactions · retention med	hanism
compounds with aromatic and multiple bond systems	no similar phases	π-π and hydrophobic	
robust C <sub>18</sub> phase for routine analyses	NUCLEOSIL <sup>®</sup> C18 Spherisorb <sup>®</sup> ODS II; Symmetry <sup>®</sup> C18; Hypersil <sup>®</sup> ODS; Inertsil <sup>®</sup> ODS II; Kromasil <sup>®</sup> C18; LiChrospher <sup>®</sup> RP-18	hydrophobic (van der Waals interactions) some residual silanol interactions	Si(CH <sub>3</sub> ) <sub>3</sub> CH <sub>3</sub> SiOH H <sub>3</sub> C N
robust C <sub>8</sub> phase for routine analyses	NUCLEOSIL® C8 ec / C8 Spherisorb® C <sub>8</sub> ; Symmetry® C <sub>8</sub> ; Hypersil® MOS; Kromasil® C <sub>8</sub> ; LiChrospher® RP-8	hydrophobic (van der Waals interactions) some residual silanol interactions	Si(CH <sub>3</sub> ) H <sub>3</sub> C $O$ Si(CH <sub>3</sub> ) H <sub>3</sub> C $O$ SiOH $\leftrightarrow$ N N O CH <sub>3</sub>
biological macromolecules like proteins or peptides	Jupiter® C4; ACE® C4	hydrophobic (van der Waals interactions) some residual silanol interactions	Si(CH <sub>3</sub> ) <sub>2</sub> SiOH ** 0 = R <sub>2</sub>
robust and well base deactivated $C_{18}$ phase; all separation tasks with preparative potential	Xterra <sup>®</sup> RP18 / MS C18 / SunFire™ C18; Luna <sup>®</sup> C18(2), Gemini <sup>®</sup> , Synergi <sup>®</sup> Max RP; Zorbax <sup>®</sup> Extend-C18; Inertsil <sup>®</sup> ODS III; Purospher <sup>®</sup> STAR RP-18; Hypersil <sup>®</sup> BDS	hydrophobic (van der Waals interactions)	Si(CH <sub>3</sub> ) <sub>3</sub> H <sub>3</sub> C
hydrophilic compounds such as polar organic acids and bases, polar natural compounds	Sequant™ ZIC <sup>®</sup> -HILIC; Obelisc™	ionic / hydrophilic and electrostatic	H <sub>3</sub> C, CH <sub>3</sub> SO <sub>3</sub> O CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> NH H <sub>3</sub> C, CH <sub>3</sub> SO <sub>3</sub> O NH <sub>2</sub> NH <sub>2</sub>
polar organic compounds (basic drugs), molecules containing π-electron systems	NUCLEOSIL <sup>®</sup> CN/CN-RP	π-π and polar (H bond), hydrophobic	
sugars, sugar alcohols and other hydroxy compounds, DNA bases, polar compounds in general	NUCLEOSIL® NH2/NH2-RP	polar / ionic and hydrophobic	OH
polar compounds in general	NUCLEOSIL® SIOH	polar / ionic	SiOH ← → O <sub>2</sub> N

### Base deactivation

NUCLEODUR<sup>®</sup> C18 Gravity and NUCLEODUR<sup>®</sup> C8 Gravity are based on the ultrapure NUCLEODUR<sup>®</sup> silica. Derivatization generates a homogeneous surface with a high density of bonded silanes (~ 18 % C for C<sub>18</sub>, ~ 11 % C for C<sub>8</sub>). Thorough endcapping suppresses any unwanted polar interactions between the silica surface and the sample, which makes "Gravity" particularly suitable for the separation of basic and other ionizable analytes. Even strongly basic pharmaceuticals like amitriptyline are eluted without tailing under isocratic conditions. For a discussion of the different retention behavior of C<sub>18</sub> phases compared to C<sub>8</sub> phases see page 40.

### Enhanced pH stability

One major disadvantage of silica stationary phases is limited stability at strongly acidic or basic pH. Cleavage of the siloxane bonding by hydrolysis, or dissolution of the silica will rapidly lead to a considerable loss in column performance. Conventional RP phases are usually not recommended to be run with mobile phases at pH > 8 or pH < 2 for extended periods of time. The special surface bonding technology and the low concentration of trace elements of NUCLEODUR<sup>®</sup> C18 and C<sub>8</sub> Gravity allow for use at an expanded pH range from pH 1 to 11.

### Benefits of enhanced pH stability

An expanded pH range is often required in method development. Many nitrogen containing compounds like basic drugs are protonated at acidic or neutral pH and exhibit poor retention on a standard  $C_{18}$  phase. The retention behavior can be improved by working at a higher pH, where the analyte is no longer protonated, but formally neutrally charged, as a rule between pH 9–10. For acidic analytes it is exactly in inverse proportion, maximum retention can be attained at low pH.

### Surface silanols at different pH values



The figure above shows the extent of protonation of surface silanols and of two exemplary analytes at acidic and alkaline pH. The following graph explains the general correlation between retention and pH.

### Correlation between retention and pH for basic and acidic compounds



An example how selectivity can be controlled by pH is the separation of the acid ketoprofen, the base lidocaine and benzamide. Under acidic conditions the protonated lidocaine is eluted very fast due to lack of sufficiently strong hydrophobic interactions

### Key features

- Suitable for LC/MS and HPLC at pH extremes (pH 1–11)
- Superior base deactivation
- Ideal for method development

### **Technical data**

- Octadecyl (C<sub>18</sub>) and octyl (C<sub>8</sub>) phase; multi-endcapped
- Pore size 110 Å; particle sizes
  1.8 µm, 3 µm and 5 µm for C<sub>18</sub>,
  1.8 µm, 3 µm and 5 µm for C<sub>6</sub>; 7 µm, 10 µm,
  12 µm and 16 µm particles for preparative purposes on request
- Carbon content 18 % for C<sub>18</sub>, 11 % for C<sub>8</sub>

#### Recommended applications

- USP listing L1
- Overall sophisticated analytical separations
- Compound classes separated include pharmaceuticals, e.g., analgesics, anti-inflammatory drugs, antidepressants; herbicides; phytopharmaceuticals; immunosuppressants

between analyte and  $C_{18}$  chains, while the formally neutral ketoprofen is eluted after about 3 min. However, at pH 10 a reversal of the elution order, with a visibly longer retention time for the basic lidocaine, is observed.



As mentioned above, pH stability of the stationary phase can be helpful for improving selectivity in method development. The following figure shows the separation of 4 basic drugs under acidic and basic conditions. At pH 2.5 the protonated analytes exhibit poor retention (early elution) and in addition an inadequate resolution for papaverine and noscapine, whilst the formally non ionized molecules can be baseline separated due to the better retention pattern at alkaline pH.



The following chromatogram demonstrates the stability of NUCLEODUR<sup>®</sup> C18 Gravity under alkaline conditions. The ultra-pure Gravity with its unique high density surface bonding technology withstands strong alkaline mobile phase conditions.



Even after 300 injections no loss of column efficiency - identified, e.g., by peak broadening or decrease in retention times – could be observed.

Under alkaline conditions dissolution of the silica support is possible, resulting in dead volume and thus peak broadening. It is worth mentioning, that this phenomenon also depends on type and concentration of buffers, as well as on the temperature. It is well known that the use of phosphate buffers, particularly at elevated temperatures, can reduce column lifetime even at moderate pH. If possible, phosphate buffers should be replaced by less harmful alternatives.

The following chromatograms show the excellent column stability of NUCLEODUR<sup>®</sup> C18 Gravity in acidic conditions. Retention times of all three compounds in the column performance test remain consistent and virtually unchanged, even after the column is run with 5000 mL eluent. Due to the extremely stable surface modification, no cleavage of the Si-O-Si bonding occurs, column deterioration is therefore successfully prevented.



# Ordering information

NUCLEODUR <sup>®</sup> C18 Gravity							
Analytical EC columns NUCLEODUR <sup>®</sup> C18 Gravity (pack of 1)							
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*			
250	4.6	5	760101.46	761903.30			
250	4	5	760101.40	761903.30			
250	3	5	760101.30	761903.30			
150	4.6	5	760103.46	761903.30			
250	3	3	760082.30	761902.30			
150	2	3	760083.20	761902.20			
125	4.6	3	760081.46	761902.30			
50	4.6	3	760080.46	761902.30			
100	4.6	1.8	760076.46	761901.30			
50	2	1.8	760079.20	761901.20			

For more products and information



\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

# Ordering information

NUCLEODUR <sup>®</sup> C8 Gravity					
Analytical EC columns NUCLEODUR <sup>®</sup> C8 Gravity (pack of 1)					
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*	
250	4.6	5	760753.46	761907.30	
250	4	5	760753.40	761907.30	
150	4.6	5	760752.46	761907.30	
150	4	5	760752.40	761907.30	
125	4.6	5	760751.46	761907.30	
125	4	5	760751.40	761907.30	
250	4.6	3	760659.46	761906.30	
50	3	3	760653.30	761906.30	
150	2	1.8	760759.20	761905.20	
50	3	1.8	760755.30	761905.30	

\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

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# Hydrophobic phase with polar selectivity

NUCLEODUR<sup>®</sup> C18 Gravity-SB excels with a relatively high hydrophobicity – similar to C18 Gravity – while simultaneously showing distinctive polar selectivity, without having polar embedded groups or polar endcapping. As a result the column displays better retention of early eluting analytes and high performance under strongly aqueous conditions. Additionally, the column is suitable for LC/MS due to low bleeding characteristics. These features are achieved through side chains (isobutyl) of the monomeric  $C_{18}$  phase.

In the TANAKA plot NUCLEODUR<sup>®</sup> C18 Gravity-SB shows similar hydrophobicity than C18 Gravity, however with a reduced capacity. The ion exchange capacity under basic conditions (pH 7.6) is high, which favors good retention of early eluting, polar substances.

Due to the broad selectivity and stability the base deactivated NUCLEODUR<sup>®</sup> C18 Gravity-SB is versatile applicable, especially for polar analytes like nucleobases or pesticides the column shows good separation efficiency.



### Key features

- Hydrophobic C<sub>18</sub> phase with distinct polar selectivity, ideal for method development, better retention of early eluting substances
- Excellent performance under highly aqueous conditions
- Suitable for LC/MS due to low bleeding characteristics

#### **Technical data**

- Monomeric octadecyl phase; extensively endcapped
- Pore size 110 Å; particle sizes 1.8 µm, 3 µm and 5 µm; carbon content 13 %; pH stability 1–9

### **Recommended applications**

- USP listing L1
- Overall sophisticated analytical separations, especially for polar compounds, e.g., antibiotics, water-soluble vitamins, organic acids



Better resolution of early eluting analyte.

# NUCLEODUR® C18 Gravity-SB

Pesticide mix (Ehrenstorfer, 17 components)					
		MN Ap	pl. No. 127330		
Column:	EC 250/4.6 NUCLEODUR®	C18 Gravity-SB, 3 µm			
Eluent:	A) acetonitrile				
	B) 5 mmol/L NH₄Ac;				
	10–37.5 % A in 50 min, 37.5	–75 % A in 25 min			15
Flow rate:	1.1 mL/min				
Temperature:	35 °C				14
Detection:	UV, 230 nm				4
Injection:	3 µL			1	5 6 1
				i i	
Peaks:					10
1. Desethylatrazir	ne 7. Chlortoluron	13. Metazachlor			$\frac{2}{3}$ 7 9 12 16
2. Metoxuron	8. Atrazine	14. Sebuthylazin			
3. Hexazinone	9. Monolinuron	15. Terbuthylazine			13 17
4. Simazine	10. Isoproturon	16. Linuron		l	
5. Cyanazine	11. Diuron	17. Metolachlor			a to and the second
6. Methabenzthia	zuron 12. Metobromuron		0 10	20	30 40 50 60 70 min

Good separation of the critical pair hexazinone/simazine



# Ordering information

NUCLEODUR <sup>®</sup> C18 Gravity-SB							
Analytical EC colur	Analytical EC columns NUCLEODUR <sup>®</sup> C18 Gravity-SB (pack of 1)						
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*			
250	4.6	5	760619.46	761992.30			
150	4.6	5	760618.46	761992.30			
150	3	5	760618.30	761992.30			
125	4	5	760617.40	761992.30			
250	3	3	760609.30	761991.30			
150	4.6	3	760608.46	761991.30			
100	2	3	760606.20	761991.20			
150	2	1.8	760598.20	761990.20			
100	2	1.8	760596.20	761990.20			
50	2	1.8	760593.20	761990.20			

\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

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# NUCLEODUR® C18 Isis

### Surface modification

By use of specific  $C_{18}$  silanes and polymeric bonding technologies a dense shield of alkyl chains protects the subjacent silica matrix. Elemental analysis of NUCLEODUR<sup>®</sup> C18 Isis shows a carbon load of 20%. The target crosslinking of the  $C_{18}$  chains on the surface enables the separation of compounds with similar molecular structure but different stereochemical properties. The technical term for this feature is steric selectivity.

### Slot model

Sander and Wise [2] proposed a model for the retention of aromatic compounds based on molecular shape, which is referred to as "Slot Model". This model pictures the bonded  $C_{18}$  phase on the silica surface with slots which analytes have to penetrate during retention. Planar molecules are able to penetrate these slots deeper than non-planar molecules of similar molecular weight and length-to-width ratio. Thus triphenylene (left structure) is retained longer than o-terphenyl (right structure).



# Steric selectivity

The following chromatograms reveal the improved resolution for positional isomers in a test mixture of aromatic compounds on NUCLEODUR<sup>®</sup> C18 Isis (green) in direct comparison with monomerically coated (blue) and polar endcapped (orange)  $C_{18}$  columns.



The separation of *o*-terphenyl and triphenylene is a good example to evaluate selectivity of a RP column in terms of the shape of two molecules. The phenyl rings of *o*-terphenyl are twisted out of plane while triphenylene has a planar geometry. The separation factor  $\alpha$  is a measure for the steric selectivity. As shown on the next page the  $\alpha$  value is considerable larger on NUCLEODUR<sup>®</sup> C18 Isis compared to a conventional C<sub>18</sub> column.

The surface bonding technology also provides improved stability features for the  $\rm NUCLEODUR^{\$}$  C18 Isis phase.

### Key features

- Phase with exceptional steric selectivity
- Outstanding surface deactivation
- Suitable for LC/MS

### **Technical data**

- C<sub>18</sub> phase with special polymeric, crosslinked surface modification; endcapped
- Pore size 110 Å; particle sizes 1.8 µm, 3 µm and 5 µm; carbon content 20 %; pH stability 1–10

### **Recommended applications**

- USP listing L1
- Steroids, (o,p,m-)substituted aromatics, fat-soluble vitamins

# NUCLEODUR® C18 Isis



### Surface deactivation

The chromatography of basic analytes requires a high density of surface-bonded  $C_{18}$  silanes combined with a thorough endcapping procedure to keep silanol activity at a minimum. This ensures tailing-free elution of even strongly basic amino-containing compounds (see application No. 121210 at ChromaAppDB.mn-net.com).

### Ordering information

NUCLEODUR <sup>®</sup> C18 Isis						
Analytical EC columns NUCLEODUR <sup>®</sup> C18 Isis (pack of 1)						
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*		
250	4.6	5	760414.46	761912.30		
250	3	5	760414.30	761912.30		
125	4	5	760412.40	761912.30		
50	3	5	760410.30	761912.30		
250	4.6	3	760404.46	761911.30		
150	4	3	760403.40	761911.30		
100	4.6	3	760401.46	761911.30		
100	4	3	760401.40	761911.30		
100	3	1.8	760407.30	761910.30		
50	4.6	1.8	760405.46	761910.30		

\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

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# RP-HPLC with highly aqueous mobile phases

The efforts to neutralize unwanted silanol activity often results in well base-deactivated RP phases with high carbon load, but a limited scope of selectivity beyond non-polar interactions. Polar compounds like carboxylic acids or drug metabolites show only weak retention on densely bonded RP columns due to distinct hydrophobic properties but low polar interactions. Very polar analytes require highly aqueous mobile phases for solubility and retention. Conventional reversed phase columns often display stability problems in eluent systems with high percentage of water (> 95 %) as evidenced by a sudden decrease of retention time and overall poor reproducibility. This phenomenon is described as phase collapse caused by the mobile phase expelled from the pores due to the fact, that hydrophobic RP phases are incompletely wetted with the mobile phase [3].

Different approaches can be used to increase column stability with highly aqueous mobile phase systems. The most promising concepts are incorporating a polar group in the hydrophobic alkyl chain, or using hydrophilic endcapping procedures to improve the wettability of the reversed phase modification. NUCLEODUR<sup>®</sup> PolarTec may be taken as an example for the embedded polar group strategy, in which a  $C_{18}$  silane with a polar function is successfully linked to the silica surface.

# Stability features

NUCLEODUR<sup>®</sup> C18 Pyramid is a silica phase with hydrophilic endcapping, designed especially for use in eluent systems of up to 100 % water. The lower figure shows the retention behavior of tartaric, acetic and maleic acid under purely aqueous conditions on NUCLEODUR<sup>®</sup> C18 Pyramid in comparison with a conventionally bonded C<sub>18</sub> phase.

It can be shown that the retention times for NUCLEODUR® C18 Pyramid remain nearly unchanged between initial injection and restart after the flow has been stopped for 12 h, whilst the performance of the conventional RP column already totally collapsed after 5 min.

### Key features

- Stable in 100 % aqueous mobile phase systems
- Interesting polar selectivity features
- Excellent base deactivation
- Suitable for LC/MS due to low bleeding characteristics

### **Technical data**

- Special C<sub>18</sub> phase; polar endcapped
- Pore size 110 Å; particle sizes 1.8 µm, 3 µm and 5 µm (7 and 10 µm particles for preparative purposes on request); carbon content 14 %; pH stability 1–9

#### **Recommended applications**

- USP listing L1
- Analgesics, penicillin antibiotics, nucleic acid bases, water-soluble vitamins, complexing agents, organic acids



### Retention characteristics

Separation of very polar compounds					
	MN Appl. No. 119170				
Column:	125 x 4 mm NUCLEODUR <sup>®</sup> C18 Pyramid, 5 µm	1	I		
Eluent:	0.2 % H <sub>3</sub> PO <sub>4</sub>				
Flow rate:	1.0 mL/min		2		
Temperature:	22 °C				
Detection:	UV, 202 nm				
Injection:	2 µL				
Peaks:		t <sub>o</sub>			
1. Formic acid		Ţ			
2. Acetic acid					
		0	2 min		

The polar surface exhibits retention characteristics different from conventional  $C_{18}$  phases. Application 119170 shows improved retention behavior of very polar short chain organic acids, which are insufficiently retained on RP columns with predominantly hydrophobic surface properties. In addition to the exceptional polar selectivity NUCLEODUR® C18 Pyramid also provides adequate hydrophobic retention (application No. 119190 at ChromaAppDB.mn-net.com). The perceptible increase in polarity has no impact on the retention behavior of ionizable analytes. Even with the strongly basic compounds of the tricyclic antidepressant drug test mixture, no unwanted interactions or a so-called lack in base deactivation are observed in application 119200.

	Tricyclic antidepressants					
MN Appl. No. 119200						
Column:	125 x 4 mm NUCLEODUR <sup>®</sup> C18 Pyramid, 5 µm	3				
Eluent:	methanol – 20 mM NH <sub>4</sub> H <sub>2</sub> PO <sub>4</sub> , pH 6.95 (70:30, v/v)					
Temperature:	40 °C	4				
Injection volume:	5 µL	2				
Detection:	UV, 254 nm					
Injection volume:	2 µL					
Peaks:		5				
1. Protriptyline	4. Imipramine					
2. Nortriptyline	5. Amitriptyline					
3. Doxepin	6. Trimipramine					

### Ordering information

NUCLEODUR <sup>®</sup> C18 Pyramid						
Analytical EC columns NUCLEODUR® C18 Pyramid (pack of 1)						
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*		
250	4.6	5	760202.46	761917.30		
250	4	5	760202.40	761917.30		
150	4.6	5	760203.46	761917.30		
125	4	5	760201.40	761917.30		
150	4.6	3	760261.46	761916.30		
125	3	3	760260.30	761916.30		
100	4.6	3	760264.46	761916.30		
50	2	3	760263.20	761916.20		
100	2	1.8	760273.20	761915.20		
50	2	1.8	760272.20	761915.20		

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# NUCLEODUR® PolarTec

# RP-HPLC under 100 % aqueous conditions

The dominant form of interactions of conventional  $C_{18}$  phases are nonpolar London dispersion forces. Besides nonpolar interactions phases with embedded polar groups possess the ability to show polar interactions (dipole-dipole, hydrogen bonds,  $\pi$ - $\pi$ , etc.). These interactions enhance retention and selectivity for polar compounds like carboxylic acids, phenols and nitrogen containing compounds.

### Key features

**Technical data** 

endcapped

1–9

Phase with embedded polar group;

**Recommended applications** 

USP listing L1 and L60

Pore size 110 Å; particle sizes 1.8 µm, 3 µm

and 5 µm; carbon content 17 %; pH stability

- RP phase with embedded polar group
- Excellent base deactivation
- Pronounced steric selectivity
- Suitable for LC/MS and 100 % aqueous eluents



In order to increase retention for polar compounds it is often necessary to decrease the organic ratio of the mobile phase to zero. Under these conditions many conventional  $C_{18}$  phases display the so-called dewetting effect which means that the mobile phase is expelled from the pores. This phenomenon leads to a dramatic loss in retention. NUCLEODUR<sup>®</sup> PolarTec is stable in 100 % aqueous mobile phases and therefore especially suited for the separation of polar compounds like organic acids.

Due to the shielding effect of the embedded group NUCLEODUR® PolarTec shows an excellent base deactivation, which is top-notch of embedded polar group phases on the market. The pronounced steric selectivity is an additional tool for the separation of complex mixtures.

Due to low bleeding characteristics NUCLEODUR® PolarTec is also suitable for LC/MS.

Even after days or weeks of operation in purely aqueous eluents the C<sub>18</sub> chains of NUCLEODUR<sup>®</sup> PolarTec are neither folded nor show any collapsing. A significant reduction of retention time cannot be observed.

# additional tool for the separation of PolarTec is also suitable for LC/MS. aqueous eluents the C<sub>18</sub> chains of show any collapsing. A significant Exceptional selectivity for phenols and nitrogen containing compounds, polar compounds like basic pharmaceuticals, organic acids, pesticides, amino acids, water-soluble vitamins, etc. Stability of NUCLEODUR<sup>®</sup> PolarTec



# NUCLEODUR® PolarTec

In spite of the polar character of the embedded functional group NUCLEODUR<sup>®</sup> PolarTec exhibits sufficient hydrophobic properties and is very well suited for analyzing basic compounds.

# Ordering information

NUCLEODUR <sup>®</sup> PolarTec					
Analytical EC colur	mns NUCLEODUR®	PolarTec (pack of 1)			
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*	
250	4.6	5	760489.46	761982.30	
250	4	5	760489.40	761982.30	
150	4.6	5	760488.46	761982.30	
150	4	5	760488.40	761982.30	
250	4.6	3	760479.46	761981.30	
150	4.6	3	760478.46	761981.30	
150	3	3	760478.30	761981.30	
150	2	1.8	760468.20	761980.20	
100	4.6	1.8	760466.46	761980.30	
50	2	1.8	760463.20	761980.20	

For more products and information Or visit www.mn-net.com



\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.



# Alternative selectivity to C<sub>18</sub> phases

Phenylhexyl modified phases are an interesting alternative to classical  $C_{18}$  phases due to an excellent separation of aromatic and unsaturated compounds especially with electron withdrawing groups.

The combination of hydrophobic and polar  $\pi$ - $\pi$  interactions result in an interesting and alternate selectivity in comparison to C<sub>18</sub> and C<sub>8</sub> modified phases.

Through short phenylhexyl chains the NUCLEODUR<sup>®</sup> Phenyl-Hexyl is more polar than the bifunctional modified NUCLEODUR<sup>®</sup> Sphinx RP. Therefor shorter analysis times can be achieved with mixtures of structural similar aromatic and aliphatic unsaturated compounds.

With NUCLEODUR<sup>®</sup> Phenyl-Hexyl e.g., tricyclic antidepressants or water soluble vitamins can be separated with good resolution.

### Key features

- Suitable for polar / aromatic compounds
- Hydrophobic phase with alternative selectivity compared to classical C<sub>18</sub> modifications
- Separation principle based on 2 retention mechanisms: π-π interactions and hydrophobic interactions
- Suitable for LC/MS due to low bleeding characteristics

### Technical data

- Phase with phenylhexyl modification; multi-endcapped
- Pore size 110 Å; particle sizes 1.8 µm, 3 µm and 5 µm; carbon content 10 %; pH stability 1–10

### **Recommended applications**

- USP listing L11
- Aromatic and unsaturated compounds, polar compounds like pharmaceuticals, antibiotics



# NUCLEODUR<sup>®</sup> Phenyl-Hexyl

Separation of water-soluble vitamins on NUCLEODUR® Phenyl-Hexyl				
	MN Appl. No. 12592	20		
Column:	100 x 3 mm NUCLEODUR <sup>®</sup> Phenyl-Hexyl, 3 µm			
Eluent:	A) 0.1 % phosphoric acid in water			
	B) 0.1 % phosphoric acid in acetonitrile			
	0 % B for 2 min, then to 60 % B in 7 min	<sup>3</sup> . 5		
Flow rate:	0.56 mL/min			
Temperature:	35 °C			
Detection:	UV, 215 nm			
Injection:	0.8 µL, 1.0 mg/mL each compound 1 mg/mL in eluent			
Peaks:				
1. Thiamine				
2. Pyridoxine		1		
3. p-aminobenzo	zoic acid			
4. Panthothenic	acid			
5. Folic acid	-			
6. Biotin	0	) 1 2 3 4 5 6 min		

# Ordering information

NUCLEODUR <sup>®</sup> Phenyl-Hexyl					
Analytical EC columns NUCLEODUR <sup>®</sup> Phenyl-Hexyl (pack of 1)					
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*	
250	4.6	5	760589.46	761987.30	
250	4	5	760589.40	761987.30	
250	3	5	760589.30	761987.30	
150	4.6	5	760588.46	761987.30	
150	4.6	3	760578.46	761986.30	
150	2	3	760578.20	761986.20	
100	4.6	3	760576.46	761986.30	
100	2	3	760576.20	761986.20	
100	3	1.8	760566.30	761985.30	
50	4.6	1.8	760563.46	761985.30	

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\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.



### Orthogonality in selectivity

Fluorinated stationary phases have been of increasing interest in HPLC. Most common representative of fluorinated silica phases is the pentafluorophenyl modification (PFP or F5). Especially the orthogonal selectivity compared to traditional alkyl phases widens the scope in analytical HPLC.

Thus NUCLEODUR<sup>®</sup> PFP offers an excellent selectivity especially for highly polar analytes like aromatic and unsaturated compounds, phenols or halogenated hydrocarbons.

While a typical C<sub>18</sub> phase just provides hydrophobic interactions between stationary phase and analyte NUCLEODUR<sup>®</sup> PFP offers four different retention mechanisms: polar interactions (H bonds), dipole-dipole,  $\pi$ - $\pi$ , and hydrophobic interactions. Especially the pronounced ion exchange capacity and distinct steric selectivity are typical for fluorinated phases.

Due to low bleeding characteristics NUCLEODUR<sup>®</sup> PFP is also suitable for LC/MS. Based on a special surface modification procedure NUCLEODUR<sup>®</sup> PFP offers highest stability also at low pH values.

NUCLEODUR<sup>®</sup> PFP offers a completely different retention behavior compared to alkyl modified silica and is often used for separations which provide insufficient results on traditional  $C_{18}$  phases.

Applications in the areas of (bio-)pharma, natural compounds and environment show the broad applicability of this phase.

### Key features

- Hydrophobic pentafluorophenyl phase with alternative selectivity in comparison to classical C<sub>18</sub> modifications
- Separation principle based on 4 retention mechanisms (polar interactions (H bonds), dipole-dipole, π-π, and hydrophobic interactions)
- Suitable for LC/MS due to low bleeding characteristics

### Technical data

- Pentafluorophenylpropyl phase; multi-endcapped
- Pore size 110 Å; particle sizes 1.8 µm, 3 µm and 5 µm; carbon content 8 %; pH stability 1–9

### **Recommended applications**

#### USP listing L43

 Aromatic and unsaturated compounds, phenols, halogen compounds, isomers, polar compounds like pharmaceuticals, antibiotics; strong retention of basic compounds





# NUCLEODUR<sup>®</sup> PFP

### Ordering information

NUCLEODUR® PFP					
Analytical EC colu	mns NUCLEODUR®	PFP (pack of 1)			
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*	
250	4.6	5	760459.46	761977.30	
250	4	5	760459.40	761977.30	
150	4.6	5	760458.46	761977.30	
125	3	5	760457.30	761977.30	
125	4	3	760447.40	761976.30	
125	3	3	760447.30	761976.30	
100	3	3	760446.30	761976.30	
100	2	1.8	760436.20	761975.20	
50	4.6	1.8	760433.46	761975.30	
50	2	1.8	760433.20	761975.20	

For more products and information



\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

# Vials and closures Optimal autosampler vials for your analysis



# Choose from

- Different vial types from N 8 to N 24 with snap ring, crimp and screw neck
- Clear glass, amber glass and polypropylene vials with or without label and scale
- Variety of closures and septa of different materials
- Diverse inserts for small sample volumes



### Highest aromatic and orthogonal selectivity

Stationary HPLC phases with biphenyl ligands like NUCLEODUR<sup>®</sup>  $\pi^2$  provide an interesting alternative to classical alkyl modified  $C_{18}$  and  $C_8$  HPLC phases due to their remarkable orthogonal selectivity.

Furthermore, the NUCLEODUR<sup>®</sup>  $\pi^2$  provides an excellent separation performance for aromatic and unsaturated analytes by combination of hydrophobic and  $\pi$ - $\pi$  interactions.

A unique feature is the predominant separation mechanism ( $\pi$ - $\pi$  or hydrophobic interactions) and thus the selectivity can be controlled by selection of the eluent. In acetonitrile/water NUCLEODUR<sup>®</sup>  $\pi^2$  shows similar retention strength to C<sub>18</sub> modified phases. Thereby displaying a significantly stronger retention than phenyl phases. These interactions are even further enhanced in a methanol/water eluent.

NUCLEODUR<sup>®</sup>  $\pi^2$  exceeds other aryl phases in terms of stability under strongly aqueous conditions. Therefore i.a. steroids, sulfonamides and acidic pharmaceuticals are separated in good resolution with NUCLEODUR<sup>®</sup>  $\pi^2$ . NUCLEODUR<sup>®</sup>  $\pi^2$  is the stationary phase with the highest aromatic analyte selectivity, which can be seen e.g., in application 127910.

### Key features

- Hydrophobic biphenylpropyl phase with alternative selectivity compared to classical C<sub>18</sub> modifications
- Separation principle based on 2 retention mechanisms (π-π interactions and hydrophobic interactions)
- Excellent performance under highly aqueous conditions

#### **Technical data**

- Biphenylpropyl phase; multi-endcapped
- Pore size 110 Å; particle size 3 μm and 5 μm; carbon content 17 %; pH stability 3–10

#### **Recommended applications**

- USP listing L11
- Overall sophisticated analytical separations, especially aromatic and unsaturated compounds, polar compounds like pharmaceuticals, antibiotics, steroids



# $\text{NUCLEODUR}^{\texttt{8}}\,\pi^2$



# Ordering information

NUCLEODUR <sup>®</sup> π <sup>2</sup>						
Analytical EC columns NUCLEODUR <sup>®</sup> π <sup>2</sup> (pack of 1)						
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*		
250	4.6	5	760625.46	761810.30		
250	4	5	760625.40	761810.30		
250	3	5	760625.30	761810.30		
250	2	5	760625.20	761810.20		
150	2	5	760624.20	761810.20		
250	4.6	3	760639.46	761811.30		
150	4.6	3	760638.46	761811.30		
150	4	3	760638.40	761811.30		
125	2	3	760637.20	761811.20		
100	3	3	760636.30	761811.30		

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\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

# Alternative RP selectivity

NUCLEODUR<sup>®</sup> Sphinx RP is characterized by exceptional selectivity features generated by a well-balanced ratio of covalently bonded octadecyl and phenyl groups. The combination of classical hydrophobic with  $\pi$ - $\pi$  interactions (aromatic ring system) expands the scope of selectivity in comparison with conventional reversed phase packings. NUCLEODUR<sup>®</sup> Sphinx RP is particularly suited for the separation of molecules containing aromatic and multiple bonds.

For the separation of polar compounds NUCLEODUR<sup>®</sup> Sphinx RP can be especially recommended and can also outperform many customary  $C_{18}$  phases. In addition, exhaustive endcapping steps minimize unwanted surface silanol activity and guarantee excellent peak shapes even for strong basic analytes.

Different from standard phenyl phases, NUCLEODUR<sup>®</sup> Sphinx RP is far more stable towards hydrolysis and is also suggested for LC/MS applications. Due to the additional intermolecular interactions NUCLEODUR<sup>®</sup> Sphinx RP is an interesting replenishment to the high density bonded phases NUCLEODUR<sup>®</sup> C8/C18 Gravity and the polar endcapped NUCLEODUR<sup>®</sup> C18 Pyramid.

### Key features

- Bifunctional RP phase with distinct selectivity based on well-balanced surface coverage
- Widens the scope for method development based on additional π-π interactions
- Suitable for LC/MS due to low bleeding characteristics

#### Technical data

- Octadecyl and propylphenyl bifunctional phase; endcapped
- Pore size 110 Å; particle sizes 1.8 µm, 3 µm and 5 µm; carbon content 15 %; pH stability 1–10

### **Recommended applications**

- USP listing L1 and L11
- Quinolone antibiotics, sulfonamides, xanthines, substituted aromatics





# NUCLEODUR<sup>®</sup> Sphinx RP

	Separation of flavonoids on t	hree different NUCLEODUR <sup>®</sup> phases			
	MN Appl. No. 119830				
Columns:	150 x 4.6 mm NUCLEODUR <sup>®</sup> Sphinx RP, 5 μm NUCLEODUR <sup>®</sup> C18 Gravity, 5 μm NUCLEODUR <sup>®</sup> C8 Gravity, 5 μm				
Eluent: Flow rate: Temperature: Detection:	water – methanol (40:60, v/v) 1 mL/min 30 °C UV, 270 nm	4 6 Sphinx RP			
Peaks: 1. Catechin 2. Rutin 3. Fisetin 4. Quercetin 5. Kaempferol 6. Isorhamnetin	3 $\mu$ L R <sub>1</sub> = R <sub>3</sub> = OH, R <sub>2</sub> = O-Rutinose R <sub>1</sub> = R <sub>2</sub> = OH, R <sub>3</sub> = H R <sub>1</sub> = R <sub>2</sub> = R <sub>3</sub> = OH R <sub>1</sub> = H, R <sub>2</sub> = R <sub>3</sub> = OH R <sub>1</sub> = OCH <sub>3</sub> , R <sub>2</sub> = R <sub>3</sub> = OH	C18 Gravity C8 Gravity			
		0.0 2.5 5.0 7.5 min			

# Ordering information

NUCLEODUR <sup>®</sup> S	NUCLEODUR <sup>®</sup> Sphinx RP					
Analytical EC col	umns NUCLEODU	IR® Sphinx RP (pack of	1)			
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*		
250	4.6	5	760803.46	761922.30		
150	4.6	5	760802.46	761922.30		
125	4	5	760801.40	761922.30		
250	4.6	3	760808.46	761921.30		
250	3	3	760808.30	761921.30		
150	3	3	760805.30	761921.30		
100	2	3	760812.20	761921.20		
100	2	1.8	760823.20	761920.20		
50	4	1.8	760822.40	761920.30		
50	3	1.8	760822.30	761920.30		

\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

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# NUCLEODUR® C18 ec for daily routine analysis

The efficiency of a separation is controlled by particle size and selectivity of the stationary phase. The exceptional surface coverage of monomeric bonded alkylsilanes, combined with an exhaustive endcapping, results in a surface with lowest silanol activity. This allows the tailing-free elution of polar compounds such as basic drugs. NUCLEODUR<sup>®</sup> C18 ec is available in 9 different particle sizes (3, 5, 7, 10, 12, 16, 20, 30 and 50  $\mu$ m) which cover the whole range from high speed analytical HPLC up to medium and low pressure prep LC. NUCLEODUR<sup>®</sup> C18 ec is also an ideal tool for scale-up purposes.

### Loading capacity

Loading capacity, probably the most important feature for preparative LC applications, is determined by pore size, pore volume and surface area of the packing. However, it can also be influenced by the molecular weight of the analytes. In the figure below the mass loading curve for acetophenone and butyrophenone on a NUCLEODUR® 100-20 C18 ec column describes the correlation between the increase of column loading and the decrease of separation efficiency.



### Key features

- Nonpolar phases for routine analysis
- Ideal and reliable standard RP phase for daily routine analysis and up-scaling for preparative HPLC
- Medium density octadecyl (C<sub>18</sub>) and octyl (C<sub>8</sub>) modification with pore size of 110 Å for a wide range of applications
- Octadecyl (C<sub>18</sub>) and butyl (C<sub>4</sub>) modification with pore size of 300 Å for the separation of biomolecules
- High batch-to-batch reproducibility

#### **Technical data**

- Medium density octadecyl, octyl and butyl phase; endcapped
- Pore size 110 Å: particle sizes 3 μm and 5 μm, 7 μm, 10 μm, 12 μm, 16 μm, 20 μm, 30 μm and 50 μm for preparative separations; carbon content 17.5% for C<sub>18</sub>, 10.5% for C<sub>8</sub>; pH stability 1–9
- Pore size 300 Å; particle size 5  $\mu$ m, carbon content 4 % for C<sub>18</sub>, 2.5 % for C<sub>4</sub>; pH stability 1–9

#### **Recommended applications**

- USP listing L1 (C<sub>18</sub>) · L7 (C<sub>8</sub>) · L26 (C<sub>4</sub>)
- 110 Å: basic, neutral or acidic drugs; derivatized amino acids; pesticides; fat-soluble vitamins; aldehydes and ketones; phenolic compounds
- 300 Å: biomolecular macromolecules, like proteins and peptides



# Chemical stability

The utmost purity of the base silica and the exceptional silane bonding chemistry minimize the risk of dissolution, or hydrolysis at pH extremes.

The chromatograms show the retention behavior at pH values of 1.5 and 10.0 for  $\rm NUCLEODUR^{\$}$  100-5 C18 ec.





# NUCLEODUR® octyl phases

In addition to NUCLEODUR<sup>®</sup> C18 phases MACHEREY-NAGEL offers octyl modified NUCLEODUR<sup>®</sup> C8 Gravity and NUCLEODUR<sup>®</sup> C8 ec columns to expand the RP tool box. Based on the same spherical high purity silica the C<sub>8</sub> phases exhibit the same chemical and mechanical stability as the C<sub>18</sub> counterparts. Indeed, NUCLEODUR<sup>®</sup> C8 Gravity can also be run at pH extremes (pH 1–11) by choosing appropriate elution parameters. Due to the shorter chain and less hydrophobic properties of the stationary phase the retention of nonpolar compounds is decreased, and in consequence a reduction in time of analysis can be achieved. Moreover, a stronger polar selectivity, particularly with the separation of ionizable analytes is frequently observed (as distinct from the C<sub>18</sub> phases). NUCLEODUR<sup>®</sup> C8 ec and NUCLEODUR<sup>®</sup> C8 Gravity are most suitable for the development of new methods but also for robust routine analyses.

There are no general guidelines which could make the choice between  $C_8$  and  $C_{18}$  phases easier but it will always be beneficial to add both phases to the existing pool of RP columns in the laboratory. Comparative studies reveal some different selectivity patterns of NUCLEODUR<sup>®</sup> C8 ec and C18 ec. The separation of phenols below shows baseline separation for 2-ethoxyphenol and dimethoxybenzene (veratrol) and in addition a reversal of the elution order of phenol and 4-methoxyphenol can be shown on the octyl phase.

### Good to know

- Octyl phases (C<sub>8</sub>) show superior polar selectivity.
- Octadecyl phases (C<sub>18</sub>) show superior hydrophobic selectivity.
- Hydrophobic compounds show shorter retention times on C<sub>8</sub> phases.



# NUCLEODUR<sup>®</sup> phases for biochromatography

A description and applications for  $C_{18}$  and  $C_4$  modified 300 Å NUCLEODUR<sup>®</sup> widepore materials for the separation of biopolymers, like peptids and proteins can be seen on the following pages.





### Sharper peaks of larger molecules on wide pore material.



Less tailing and better separation on NUCLEODUR® 300-5 C18 ec.

# Ordering information

NUCLEODUR <sup>®</sup> C18 ec				
Analytical EC colu	mns NUCLEODUR®	C18 ec (pack of 1)		
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*
250	4.6	5	760002.46	761932.30
250	4	5	760002.40	761932.30
150	4.6	5	760008.46	761932.30
125	4.6	5	760001.46	761932.30
125	4	5	760001.40	761932.30
125	3	5	760001.30	761932.30
125	2	5	760001.20	761932.20
250	4.6	3	760052.46	761931.30
250	4	3	760052.40	761931.30
150	4.6	3	760053.46	761931.30

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\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

# Ordering information

NUCLEODUR® C1	NUCLEODUR® C18 ec					
Preparative VP co	lumns NUCLEODUF	R <sup>®</sup> C18 ec (pack of 1)	)			
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*		
250	21	10	762010.210	762090.160		
250	10	10	762010.100	762090.80		
250	21	5	762022.210	762090.160		
250	10	5	762022.100	762090.80		
50	10	5	762003.100	762090.80		

\* For more information of guard columns for preparative VP columns please see page 91.

# Ordering information

NUCLEODUR <sup>®</sup> 300-5 C18 ec				
Analytical EC colur	mns NUCLEODUR <sup>®</sup>	300-5 C18 ec (pack	of 1)	
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*
250	4.6	5	760186.46	761988.30
250	4	5	760186.40	761988.30
150	4.6	5	760185.46	761988.30
150	2	5	760185.20	761988.20
100	4.6	5	760183.46	761988.30

\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

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# Ordering information

NUCLEODUR <sup>®</sup> C8 ec				
Analytical EC colu	mns NUCLEODUR®	C8 ec (pack of 1)		
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*
250	4.6	5	760703.46	761937.30
250	4	5	760703.40	761937.30
150	4.6	5	760702.46	761937.30
125	4	5	760701.40	761937.30
50	4.6	5	760700.46	761937.30
100	3	5	760704.30	761937.30
250	4	3	760062.40	761936.30
150	4.6	3	760061.46	761936.30
125	4.6	3	760060.46	761936.30
125	2	3	760060.20	761936.20

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\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

# Ordering information

NUCLEODUR® C8 ec					
Preparative VP columns NUCLEODUR <sup>®</sup> C8 ec (pack of 1)					
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*	
250	21	5	762062.210	762092.160	
250	10	5	762062.100	762092.80	
125	10	5	762061.100	762092.80	

\* For more information of guard columns for preparative VP columns please see page 91.

# Ordering information

NUCLEODUR® 300-5 C4 ec				
Analytical EC colur	mns NUCLEODUR®	300-5 C4 ec (pack o	f 1)	
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*
250	4.6	5	760196.46	761989.30
150	4.6	5	760195.46	761989.30
100	4.6	5	760193.46	761989.30
100	4	5	760193.40	761989.30
100	2	5	760193.20	761989.20

\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

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### The preparative octadecyl phase

Preparative separations place high demands on silica based HPLC materials. Apart from excellent selectivity and base deactivation, robustness (pH, pressure stability, ...) and capacity are vital criteria for optimal and efficient separation at the preparative scale.

### Selectivity and base deactivation

The innovative endcapping procedure leads to exceptionally good base deactivation – the Engelhardt test demonstrates superb selectivity, peak symmetry and peak shape over the entire polarity range. In addition, NUCLEODUR<sup>®</sup> C18 HTec scores in low bleed characteristics and is therefore highly suitable for LC/MS.

### Key features

- Base-deactivated preparative octadecyl phase
- Reliable and durable standard RP phase for up-scaling to preparative scale
- High loading capacity and excellent stability
- Suitable for LC/MS

#### **Technical data**

- High density octadecyl (C<sub>18</sub>) phase; multi-endcapped
- Pore size 110 Å; particle sizes 1.8 μm, 3 μm, 5 μm, 7 μm and 10 μm for analytical and preparative separations; carbon content 18%, pH stability 1–11



# Stability and lifetime

Based on fully synthetic and extremely robust totally spherical NUCLEODUR<sup>®</sup> silica, NUCLEODUR<sup>®</sup> C18 HTec offers outstanding mechanical rigidity and thus the perfect choice for self-packing of prep-columns, too. The special surface modification and endcapping procedure results in high chemical stability even at extreme chromatographic conditions like high flow rates, high temperature or critical solvents (DMSO). Furthermore, NUCLEODUR<sup>®</sup> C18 HTec columns show a remarkably long lifetime in acidic (pH 1) as well as basic (pH 10) mobile phases.

### **Recommended applications**

USP listing L1

 Sophisticated analytical and preparative separations of basic, neutral and acidic pharmaceuticals, derivatized amino acids, pesticides, fat-soluble vitamins, aldehydes, ketones and phenolic compounds



# NUCLEODUR® C18 HTec

### Up-scaling

Due to highest quality standards in silica production and phase chemistry combined with optimized packing technology, NUCLEODUR<sup>®</sup> C18 HTec allows exceptional transferability from analytical to preparative scale with respect to different particle sizes (e.g., 5, 7 or 10  $\mu$ m) as well as column dimensions (e.g., ID 4.6 to 21 mm).

### Good to know

 Due to innovative surface coating procedures NUCLEODUR<sup>®</sup> C18 HTec offers excellent analytical separation properties and is the first choice for up-scaling to preparative column dimensions.



# Capacity

A vital criterion for efficiency in preparative HPLC is the capacity of the separation medium. NUCLEODUR® C18 HTec is characterized by a notably high loading capacity under both basic and acidic conditions, while competitor columns show overload effects even at lower loadings (×).

	Loading capacity under acidic conditions	
	MN Appl. No. 123890	
Columns:	VP 100 x 21 mm NUCLEODUR <sup>®</sup> C18 HTec, 5 µm	
	100 x 21.2 mm AXIA™ Gemini <sup>®</sup> 5 µm C18 110 Å	
Eluent:	acetonitrile – formic acid in $H_2O pH 3.0$	
	(30:70, v/v)	
Flow rate:	28 mL/min	0
Temperature:	22 °C	2
Pressure:	124 bar	1
Detection:	UV, 254 nm	
		3
Peaks:		
total load 40 mg		
(sample dissolve	d in DMSO)	$\wedge    \times   $
1. 4-Acetamidop	henol (5 mg)	
2. 2-Acetamidophenol (10 mg)		-
3. Acetylsalicylic	acid (25 mg)	0 1 2 min
L		



# NUCLEODUR® C18 HTec

# Ordering information

NUCLEODUR <sup>®</sup> C18 HTec				
Analytical EC colu	mns NUCLEODUR®	C18 HTec (pack of 1	)	
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*
250	4.6	5	760316.46	761927.30
250	4	5	760316.40	761927.30
150	4.6	5	760315.46	761927.30
125	4	5	760314.40	761927.30
250	4.6	3	760326.46	761926.30
150	4.6	3	760325.46	761926.30
150	2	3	760325.20	761926.20
125	3	3	760324.30	761926.30
150	2	1.8	760308.20	761925.20
100	2	1.8	760306.20	761925.20

For more products and information



\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

# Ordering information

NUCLEODUR <sup>®</sup> C18 HTec				
Preparative VP col	umns NUCLEODUR	<sup>®</sup> C18 HTec (pack of	1)	
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*
250	50	10	762576.500	762592.500
250	21	10	762576.210	762591.160
250	10	10	762576.100	762591.80
150	32	10	762575.320	762592.320
250	21	7	762566.210	762591.160
250	10	7	762566.100	762591.80
250	50	5	762556.500	762592.500
250	40	5	762556.400	762592.320
250	32	5	762556.320	762592.320
250	21	5	762556.210	762591.160
250	16	5	762556.160	762591.160
250	10	5	762556.100	762591.80
250	8	5	762556.80	762591.80
150	32	5	762555.320	762592.320
100	21	5	762553.210	762591.160

\* For more information of guard columns for preparative VP columns please see page 91.

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# NUCLEODUR<sup>®</sup> HILIC

### Hydrophilic interaction chromatography



Especially for polar compounds reversed phase HPLC – the most common analytical method – is often limited. Here, hydrophilic stationary phases provide an additional tool for the separation of polar analytes in HPLC.

The expression HILIC (Hydrophilic Interaction Chromatography) was firstly published by Andrew Alpert in 1990 – since then it took quite some efforts to develop robust and reproducible hydrophilic HPLC phases for HILIC chromatography [4].

HILIC combines the characteristics of the 3 major methods in liquid chromatography reversed phase (RPC), normal phase (NPC) and ion chromatography (IC):

- $\cdot$  Stationary phases (adsorbents) are mostly polar modifications of silica or polymers (SiOH, NH<sub>2</sub>, Diol, (zwitter) ions, ...) like in NPC.
- Mobile phases (eluents) are mixtures of aqueous buffer systems and organic modifiers like acetonitrile or methanol – like in RPC.
- Fields of application include quite polar compounds as well as organic and inorganic ions – like in IC.

Summarized: "HILIC is NP chromatography of polar and ionic compounds under RP conditions."



NUCLEODUR<sup>®</sup> HILIC is a special zwitterionic modified stationary phase based on ultra-spherical NUCLEODUR<sup>®</sup> particles. The betaine character of the ammonium-sulfonic acid ligands results in total charge equalization and in an overall neutrally but highly polar surface.

### Retention characteristic

Commonly HILIC is described as partition chromatography or liquid-liquid extraction system between mobile and stationary phases. Versus a water-poor mobile phase a water-rich layer on the surface of the polar stationary phase is formed. Thus, a distribution of the analytes between these two layers will occur. Furthermore, HILIC includes weak electrostatic mechanisms as well as hydrogen donor interactions between neutral polar molecules under high organic elution conditions. This distinguishes HILIC from ion exchange chromatography - main principle for HILIC separation is based on compound's polarity and degree of solvation.

### Stability features

Due to an advanced and unique surface modification procedure NUCLEODUR<sup>®</sup> HILIC columns provide short equilibration times. After just 20 min equilibration the 2nd injection already shows stable and reproducible results.

### Key features

- Ideal for reproducible and stable chromatography of highly polar analytes
- Suitable for analytical and preparative applications
- Very short column conditioning period

#### **Technical data**

- Zwitterionic ammonium-sulfonic acid phase; not endcapped
- Pore size 110 Å; particle sizes 1.8 µm, 3 µm and 5 µm; carbon content 7 %; pH stability 2–8.5

### **Recommended applications**

 Hydrophilic compounds such as organic polar acids and bases, polar natural compounds, nucleosides, oligonucleotides, amino acids, peptides, water soluble vitamins

# NUCLEODUR<sup>®</sup> HILIC

Beyond this, NUCLEODUR<sup>®</sup> HILIC columns are characterized by an outstanding column life time - even after nearly 800 runs the columns show no loss of its pristine performance - peak shape and retention are still immaculate. Due to its high loading capacity NUCLEODUR<sup>®</sup> HILIC is suitable for (semi-)preparative applications.

### Good to know

NUCLEODUR<sup>®</sup> HILIC is a patented phase modification (pat. number DE102009006007 (B4))



Overall NUCLEODUR<sup>®</sup> HILIC provides excellent chromatographic features and is hereby the perfect choice for separation of polar or charged compounds which can be shown in application 122920.

	Separation of adenosine and phosphates					
	MN Appl. No. 122920					
Column:	125 x 4 mm NUCLEODUR <sup>®</sup> HILIC, 5 μm	1				
Eluent:	acetonitrile – 100 mM ammonium acetate, pH 5.3 (70:30, v/v)					
Flow rate:	1.3 mL/min					
Temperature:	25 °C					
Detection:	UV, 254 nm					
Peaks:		2				
1. Adenosine	4. ADP	3 4				
2. cAMP	5. ATP					
3. AMP						

# Ordering information

NUCLEODUR <sup>®</sup> HILIC					
Analytical EC columns NUCLEODUR <sup>®</sup> HILIC (pack of 1)					
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*	
250	4.6	5	760550.46	761962.30	
150	2	5	760553.20	761962.20	
250	4.6	3	760530.46	761961.30	
250	3	3	760530.30	761961.30	
150	4.6	3	760533.46	761961.30	
125	4.6	3	760531.46	761961.30	
125	2	3	760531.20	761961.20	
100	3	3	760534.30	761961.30	
100	2	1.8	760526.20	761960.20	
50	2	1.8	760523.20	761960.20	

\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

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### Alternative bonded-phase functionality

In reversed phase HPLC it is fairly common to start with  $C_{18}$  or  $C_8$  columns when developing new methods. However, superior polarity and selectivity properties often required for more sophisticated separations, are not always sufficiently provided by classical RP phases. These classical RP phases are usually characterized by a hydrophobic layer of monomeric or polymeric bonded alkylsilanes.

One approach to improve the resolution of compounds poorly separated on nonpolar stationary phases is to change bonded-phase functionality.

The fully endcapped and highly reproducible NUCLEODUR<sup>®</sup> 100-5 CN-RP phase has cyanopropyl groups on the surface able to generate clearly recognizable and different retention behavior compared to purely alkyl-functionalized surface modifications (see application 119340).

The polarity of NUCLEODUR<sup>®</sup> 100-5 CN-RP can be classified as intermediate based on multiple retention mechanisms such as dipole-dipole,  $\pi$ - $\pi$ , and also hydrophobic interactions [5]. Therefore, this phase shows a distinct selectivity for polar organic compounds as well as for molecules containing  $\pi$  electron systems (e.g., analytes with double bonds, tricyclic antidepressants) [6].

Short-chain bonded phases sometimes reveal shortcomings in stability towards hydrolysis at low pH [7]. Application 119350 shows that even after 100 sample injections and four weeks storage at pH 1 (blue curve), neither a considerable shift in retention, nor a visible change in peak symmetry could be noticed (green curve = new column).

### Key features

- Multi-mode phase modification (RP and NP) widens scope of selectivity
- High retention capacity especially for very polar and unsaturated compounds
- Stable against hydrolysis at low pH
- $\hfill \label{eq:comparison}$  Different retention characteristics in comparison to C\_8 and C\_{18} phases

#### **Technical data**

- Cyanopropyl high purity phase; specially endcapped
- Pore size 110 Å; particle sizes 3 µm and 5 µm; carbon content 7 %; pH stability 1–8

#### Recommended applications

- USP listing L10
- Tricyclic antidepressants, steroids, organic acids





# NUCLEODUR® CN/CN-RP

### Multi-mode columns

Due to its polarity, the cyano phase can also be run in normal phase mode. NUCLEODUR<sup>®</sup> CN columns for NP applications are shipped in *n*-heptane. The change in selectivity and order of elution for a mixture of various steroids in NP and RP mode is displayed below. The high coverage combined with a thorough endcapping makes NUCLEODUR<sup>®</sup> 100-5 CN-RP suitable for separation of ionizable compounds such as some of the basic drugs analyzed in applications 119271 and 119272.



### Ordering information

NUCLEODUR <sup>®</sup> CN-RP/CN					
Analytical EC columns NUCLEODUR <sup>®</sup> CN-RP (pack of 1)					
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*	
250	4.6	5	760152.46	761944.30	
250	4	5	760152.40	761944.30	
150	4.6	5	760154.46	761944.30	
125	4.6	5	760153.46	761944.30	
150	4.6	3	760156.46	761941.30	
150	4	3	760156.40	761941.30	
50	2	3	760159.20	761941.20	
Analytical EC columns NUCLEODUR® CN (pack of 1)					
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*	
250	4.6	5	760150.46	761943.30	
250	4	5	760150.40	761943.30	
150	4.6	5	760149.46	761943.30	

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\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

### Amino modified HPLC phase

Some compounds, especially polar substances, cannot be sufficiently resolved on  $C_{18}$  phases. Polar-modified silica phases offer alternative selectivities thus expanding the spectrum of analytical HPLC into the polar range.

### Multi-mode columns

Besides cyano modifications, amino modifications belong to the most frequently used polar silica phases. They both feature an important advantage – that they can be run in the RP and NP mode. RP mode using aqueous-organic eluent mixtures and NP mode with hexane as a possible mobile phase.

NUCLEODUR<sup>®</sup> NH2, belongs to the so-called multi-mode columns. It can be used for RP chromatography of polar compounds (such as sugars in aqueous-organic eluent systems), for NP chromatography of substituted aromatics or chlorinated pesticides with organic mobile phases (such as hexane, dichloromethane or 2-propanol), but also for ion exchange chromatography of anions and organic acids using conventional buffers and organic modifiers.

The main field of application of NUCLEODUR<sup>®</sup> NH2 is the separation of simple and complex sugars, sugar alcohols, and other hydroxy compounds under RP conditions as well as hydrocarbons under NP conditions.



### Key features

- Multi-mode phase modification (for RP, NP and IC)
- Stable against hydrolysis at low pH and stable in 100 % aqueous eluents
- Widens scope of analytical HPLC into the polar range
- Suitable for LC/MS

#### **Technical data**

- Aminopropyl high purity phase; not endcapped
- Pore size 110 Å; particle sizes 3 µm, 5 µm and 7 µm; carbon content 2.5 %; pH stability 2–8

### **Recommended applications**

- USP listing L8
- Polar compounds under RP conditions (sugars, DNA bases), hydrocarbons under NP conditions



# NUCLEODUR® NH2/NH2-RP

Due to the special method of surface modification NUCLEODUR<sup>®</sup> NH2 features a pronounced stability at higher as well as lower pH values. The following figure shows, that even after several days of exposure of the column material at pH 1.75 good separation efficiency and peak symmetry are maintained. The resulting high column life allows cost reduction due to lower column consumption.

The following example shows enhanced pH stability of NUCLEODUR<sup>®</sup> NH2 and outstanding suitability for the separation of total herbicides (AMPA, glyphosate, glufonisate, ...) - see application No. 122190 in our online database at ChromaAppDB.mn-net.com.

#### Good to know

- Normal phase chromatography (NP) with hexane, dichloromethane or 2-propanol as mobile phase for polar compounds
- Reversed phase chromatography (RP) of polar compounds in aqueousorganic eluent systems
- Ion exchange chromatography of anions and organic acids using conventional buffers and organic modifiers

min

Based on spherical NUCLEODUR<sup>®</sup> silica this phase features a high pressure stability which makes it the perfect choice for preparative separations as well as for LC/MS. Additionally, the high batch-to-batch reproducibility of NUCLEODUR<sup>®</sup> NH2 enables reliable analyses especially for routine work.

### Ordering information

NUCLEODUR® NH2-RP/NH2					
Analytical EC columns NUCLEODUR <sup>®</sup> NH2-RP (pack of 1)					
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*	
250	4.6	5	760732.46	761953.30	
250	4	5	760732.40	761953.30	
250	2	5	760732.20	761953.20	
125	4	5	760730.40	761953.30	
250	4.6	3	760739.46	761951.30	
150	4.6	3	760742.46	761951.30	
100	2	3	760740.20	761951.20	
Analytical EC columns NUCLEODUR <sup>®</sup> NH2 (pack of 1)					
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*	
250	4.6	5	760722.46	761952.30	
250	4	5	760722.40	761952.30	
125	4.6	5	760720.46	761952.30	

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\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

# Unmodified ultra-pure spherical silica gel

Unmodified silica gels were the first substrates used for liquid chromatography. Our ultra-pure spherical silica gel NUCLEODUR<sup>®</sup> SiOH can be used under normal phase as well as HILIC conditions. With no phase modification and no endcapping, the bare silica phase is suited for applications with polar to midpolar compounds. Due to this, very polar silica surface with silanol and siloxane groups a non-polar mobile phase is necessary for a perfect chromatographic performance. NUCLEODUR<sup>®</sup> SiOH can be used for analytical and preparative HPLC applications.



# Ordering information

NUCLEODUR <sup>®</sup> SiOH					
Analytical EC columns NUCLEODUR <sup>®</sup> SiOH (pack of 1)					
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*	
250	4.6	5	760007.46	761967.30	
250	4	5	760007.40	761967.30	
150	4.6	5	760012.46	761967.30	
250	4.6	3	760173.46	761966.30	
150	4.6	3	760172.46	761966.30	

\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

### Key features

- Unmodified silica for NP chromatography
- Totally spherical high purity silica
- Pressure stable up to 600 bar
- Suitable for analytical and preparative separation of polar and midpolar compounds

### **Technical data**

- Unmodified high purity phase; not endcapped
- Pore size 110 Å; particle sizes 3 to 50 μm; pore volume 0.9 mL/g; surface area (BET) 340 m<sup>2</sup>/g; pH stability 2–8; metal content
   < 10 ppm (see page 10)</li>

#### Recommended applications

- USP listing L3
- Polar and midpolar compounds under normal phase and HILIC conditions

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# CHROMABOND<sup>®</sup> SPE products For a wide variety of applications



# High-performance products for sample preparation

- Comprehensive range of RP and normal phases as well as ion exchangers
- Polymer and silica based phases
- Phases for special applications like food or environmental analysis



# NUCLEODUR® C18 PAH

### Analysis of polycyclic aromatic hydrocarbons (PAHs) by HPLC

Polycyclic aromatic hydrocarbons (PAHs) are chemical compounds that consist of fused aromatic rings and do not contain heteroatoms or carry substituents. As a pollutant, they are of concern because some compounds have been identified as carcinogenic, mutagenic, and teratogenic. PAHs are natural components of coal or gas. They are delivered to our environment by pyrolysis (incomplete burning) of organic materials like coal, oil, fuel, wood, and tabacco; hence it can found globally. Today most PAHs accrue from anthropogenic processes – but also natural origins such as forest fire. In the past the production of coke and gas from black coal had a considerable impact on environmental pollution. Waste products (e.g., tar) from coking or gas plants are often the origin of serious ground water pollutions.

Since a number of PAHs (e.g., benzo[a]pyrene, 3-methylcholanthrene and benzanthracene) have been proven to be carcinogenic. Therefore control of the PAH content in food, water, and soil is an important task for routine analysis. For choice and limiting values of the polycyclics we refer to the governmental regulations, which exist in many countries (e.g., EPA method 610 of the United States Environmental Protection Agency).

PAHs can be determined by different chromatographic techniques (TLC, GC, HPLC). Thus the 6 PAHs according to German drinking water specification (TVO) can, e.g., be analyzed by TLC (see German Standard DIN 38 409), while a much larger number of polycyclic aromatics can be determined by GC or HPLC.

#### Key features

- Special octadecyl phase for PAH analysis
- Base material high purity NUCLEODUR<sup>®</sup> silica

### **Technical data**

- Special octadecyl phase with polymerically coated base material; endcapped
- Pore size 110 Å, particle sizes 1.8 μm and 3 μm

#### **Recommended applications**

- USP listing L1
- Allows efficient gradient separation of the 16 PAHs according to EPA



Detection of separated PAHs with UV (250–280 nm), diode array or fluorescence detection at different wavelengths for excitation and emission (acenaphthylene cannot be analyzed with fluorescence detection).

# NUCLEODUR<sup>®</sup> C18 PAH

		Separation of 18 PAHs on N	UCLEODUR <sup>®</sup> C18 PAH		
		MN Appl. No.	123840		
Column:	125 x 4 mm				
	NUCLEODUR <sup>®</sup> C18 PA	H, 3 μm			
Eluent:	A) methanol – water				
	(70:30, v/v); B) acetonit	rile		Benzo[a]anthrace	en Benzo[a]pyren
	0–20 % B in 1.5 min,				
	20–50 % B in 1.5 min,				
	50–100 % B in 1.0 min			3 6	
	100 % B for 3 min,				
	100–0 % B in 0.5 min				
Flow rate:	1.5 mL/min			5	
Temperature:	35 °C			7	
Injection:	UV: 1 μL,			8 10	) 12
Fluorescence:	0.5 µL		0	9	11 10
Detection:	UV, 254 nm		2-me- 1-me-r	-n	
	fluorescence				
	(see chromatogram)			4	14 15
Peaks:				11 î	13
1. Naphthalene		10. Chrysene	1		
2. Acenaphthyle	ene (not detectable by	11. Benzo[b]fluoranthene			
fluorescence)		12. Benzo[k]fluoranthene			
3. Acenaphthene		13. Benzo[a]pyrene	0		
4. Fluorene		14. Dibenz[ah]anthracene			
5. Phenantrene		15. Benzo[ghi]perylene			
6. Anthracene		16. Indeno[1,2,3-cd]pyrene			
7. Fluoranthene		1-me-n: 1-methylnaphthalene	λ <sub>ex</sub> 2	75 375 335 315	330 375 345 300 nm
8. Pyrene		2-me-n: 2-methylnaphthalene	Λ <sub>em</sub> 3	425 440 405	420 400 420 500 nm
9. Benz[a]anthra	acene		0 1 2	3 4	5 6 7 min



### HPLC columns for PAH analysis

For PAH analyses we have developed a specially modified  $C_{18}$  phase based on NUCLEODUR<sup>®</sup> which allows efficient gradient separation of 16 PAHs according to EPA regulations. Detection of the separated PAHs can be achieved by UV (250–280 nm), with diode array or with fluorescence detection at different wavelengths for excitation and emission. Acenaphthylene cannot be analyzed with fluorescence detection. For cost-effective routine PAH analysis we recommend applications using methanol instead of acetonitrile as the eluent. For rapid analysis NUCLEODUR<sup>®</sup> C18 PAH (3  $\mu$ m) in short columns (100 mm) provides excellent results at high flow rates. Hereby separation of 16 PAHs according to EPA can be achieved in less than 3 min.

Tightened regulations require determination of 2 additional PAHs (1- and 2-methylnaphthalene) – so we developed highly efficient methods for 18 PAHs on the NUCLEODUR<sup>®</sup> C18 PAH.

# Ordering information

NUCLEODUR <sup>®</sup> C18 PAH					
Analytical EC columns NUCLEODUR <sup>®</sup> C18 PAH (pack of 1)					
Length (mm)	ID (mm)	Particle size (µm)	REF	Guard columns*	
250	4	3	760786.40	761971.30	
250	3	3	760786.30	761971.30	
150	3	3	760785.30	761971.30	
125	4	3	760784.40	761971.30	
125	3	3	760784.30	761971.30	
100	3	3	760783.30	761971.30	
100	4	1.8	760773.40	761970.30	
100	3	1.8	760773.30	761970.30	
100	2	1.8	760773.20	761970.20	

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\* Pack of 3, EC guard columns require column protection system REF 718966. For more information, see page 90.

Ako nás možno kontaktovať:

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